

## TWO IRIDOIDS FROM *VIBURNUM LANTANA*

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**Key Word Index**—*Viburnum lantana*; Caprifoliaceae; iridoids; 2'-O-acetyldihydropenstemide; 2'-O-acetylpatrinoside.

**Abstract**—Two new iridoids with an isovaleroyl group at C-1 and a sugar moiety at C-11 have been isolated from the bark of *Viburnum lantana*. Their structures were characterized as 2'-O-acetyldihydropenstemide and 2'-O-acetylpatrinoside by spectroscopic and chemical means.

### INTRODUCTION

*Viburnum* species are used in folk and official medicine as uterotonic, chemostatic, sedative and diuretic drugs [1]. They are a source of iridoid compounds [2–10], two new ones of which we have isolated from the previously unstudied species *V. lantana*.

### RESULTS AND DISCUSSION

Compounds **1** and **2** were isolated from the dried bark of *V. lantana*. Both compounds after heating with dilute hydrochloric acid produced a dark resinous product, typical for iridoids, and afforded glucose [TLC and GC (silylated derivative)].

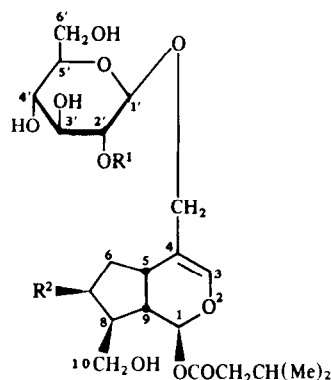
Compound **1** exhibited IR absorption bands at 3450 (hydroxyl groups), 1730–1750 (ester group), 1670 (double bond) and 1250 cm<sup>-1</sup> (acetoxy group). Its <sup>1</sup>H and <sup>13</sup>C NMR spectra were similar to those of dihydropenstemide (**3**) [4,11] with the exception of the signals for one acetoxy group, indicating that **1** was a monoacetate of **3** (Table 1).

Acetylation of **1** provided a penta-acetate (**1a**), the *M<sub>r</sub>* mass (656) of which as determined by CIMS (Et<sub>2</sub>NH) [12, 13] showed **1** to have a *M<sub>r</sub>* of 488. The presence of two intense peaks at *m/z* 422 (60%) [gluAc<sub>4</sub> + 74]<sup>+</sup> and at 298 (50%) [M + Et<sub>2</sub>NH<sub>2</sub> - 102 - 330]<sup>+</sup> characterized the sugar and the aglycone parts (Scheme 1).

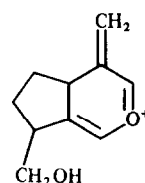
Furthermore, comparison of the <sup>13</sup>C NMR spectra of **1a** and dihydropenstemide penta-acetate **3a** showed their common identity (Table 2). The site of esterification was established by comparison of the <sup>1</sup>H and <sup>13</sup>C NMR spectra of **1** with those of **3**. The <sup>1</sup>H NMR spectrum of **1** indicated that the acetoxy residue was linked to the glucose moiety. No shift of the methylene protons at C-10 (δ 3.53) was observed while the H-2' signal showed a considerable paramagnetic shift (δ 4.70 cf δ 3.18 for **3**)

(Table 1). The C-2' position of the acetoxy group was supported also by the low field shifts of the C-1' and C-3' signals in the <sup>13</sup>C NMR spectrum of **1** compared to those of dihydropenstemide (**3**).

The EIMS (12 eV) of **1** was in agreement with the proposed structure of the aglycone. Signals with *m/z* 165 and 164 being due to the aglycone of **1** after elimination of isovaleric acid and either water or a hydroxyl group. The



	R <sup>1</sup>	R <sup>2</sup>
<b>1</b>	Ac	H
<b>2</b>	Ac	OH
<b>3</b>	H	H
<b>4</b>	H	OH

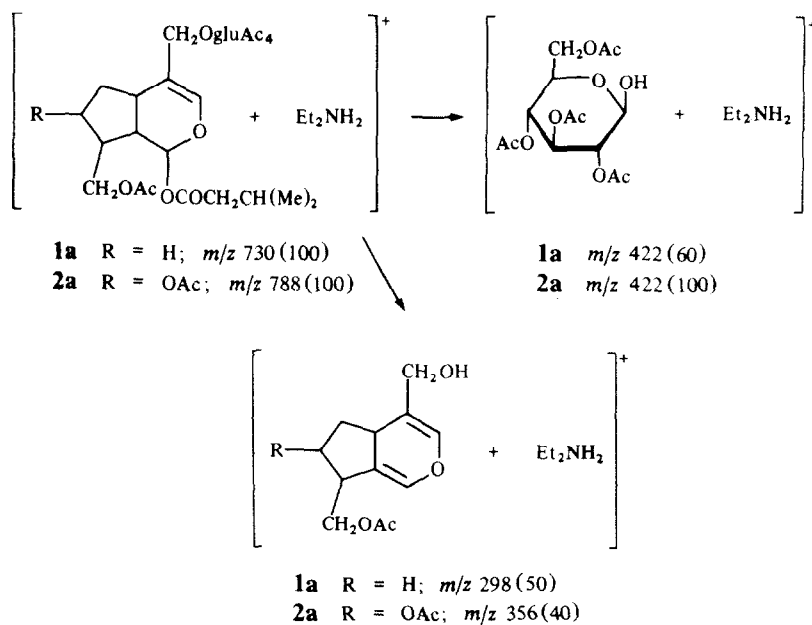


**5**

*m/z* 165(85)

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Scheme 1. CIMS ( $\text{Et}_2\text{NH}$ ) fragmentation of compounds **1a** and **2a**.Table 1.  $^1\text{H}$  NMR data of

H	<b>1</b> *	<b>3</b> †	<b>1a</b> ‡
1	5.99 <i>d</i> (4.5)	5.96 <i>d</i> (4.6)	5.94 <i>d</i> (5.5)
3	6.32 <i>br s</i>	6.37 <i>br s</i>	6.32 <i>d</i> (1.5)
5	2.70 <i>m q-habitus</i>	2.82 <i>m q-habitus</i>	2.67 <i>m q-habitus</i>
6	1.82 <i>m</i>	1.39 <i>m</i>	1.6 <i>m</i>
	1.69 <i>m</i>	1.70 <i>m</i>	1.7–2.1 <i>m</i>
7	1.9–2.2 <i>m</i>	1.90–2.02 <i>m</i>	1.7–2.1 <i>m</i>
	1.40 <i>m</i>	1.82 <i>m</i>	1.34 <i>m</i>
8	1.9–2.2 <i>m</i>	1.90–2.02 <i>m</i>	2.17 <i>m</i>
9	1.9–2.2 <i>m</i>	1.90–2.02 <i>m</i>	2.10 <i>m</i>
10	3.53 <i>d</i> (6.0)	3.52 <i>d</i> (6.0)	4.04 <i>d</i> (7.0)
11	4.21 <i>d</i> AB (12.0)	4.16 <i>d</i> AB centre	4.04 <i>d</i> AB (12.0)
	4.04 <i>d</i> AB (12.0)	(11.5)	4.20 <i>d</i> AB (12.0)
1'	4.46 <i>d</i> (8.0)	4.28 <i>d</i> (7.7)	4.54 <i>d</i> (8.0)
2'	4.70 <i>dd</i> (8.0; 9.0)	3.18 <i>dd</i> (7.8; 9.1)	5.02 <i>dd</i> (8.0; 9.5)
3'	3.53 <i>t</i> (9.0)		5.22 <i>t</i> (9.5)
4'	3.35		5.08 <i>t</i> (9.5)
5'	3.28		3.69 <i>m</i>
6'	3.88 <i>dd</i> AB from	3.86 AB from ABX	4.26 <i>dd</i> AB from
	ABX (12.0; 2.2)	(11.7; 1.8)	ABX (11.5; 5.0)
	3.68 <i>dd</i> AB from	3.65 AB from ABX	4.14 <i>dd</i> AB from
	ABX (12.0; 5.5)	(11.7; 5.4)	ABX (11.5; 2.5)
MeCO-	2.1 <i>s</i>		
isovaleroyl			
–CH<	1.9–2.2 <i>m</i>	2.07	1.7–2.1 <i>m</i>
–CH <sub>2</sub> –	2.23 <i>d</i>	2.22	2.25 <i>d</i>
(CH <sub>3</sub> ) <sub>2</sub>	0.96 <i>d</i>	0.96 <i>d</i>	0.98 <i>d</i>

\***1** and **2** in  $\text{CD}_3\text{OD}$ , 250 MHz.†**3** in  $\text{CD}_3\text{OD}$ , 400 MHz, ref. [11].‡**1a** and **2a** in  $\text{CDCl}_3$ , 250 MHz.§**4a** in  $\text{CDCl}_3$ , 100 Hz, ref. [15].

||Partially covered by the solvent signal.

presence of the isovaleroyl and acetyl moieties in the molecule was shown by the intense peaks at  $m/z$  85, 57, 60 and 43 in the EIMS (70 eV) of **1**.

The stereochemistry at C-8 of compound **1** (C-9,  $\delta$ 45.0; C-10, 66.6; H-1, 5.96) was studied by  $^{13}\text{C}$  NMR [14] and by comparison of the  $^{13}\text{C}$  and  $^1\text{H}$  NMR data with those of dihydropenstemide **3** (C-9,  $\delta$ 44.95; C-10, 66.48; H-1, 5.96) [11] and 8-epidihydropenstemide (C-9,  $\delta$ 42.71; C-10, 64.29; H-1, 6.26) [11]. All the data were consistent with an 8- $\beta$ - $\text{CH}_2\text{OH}$  substituent. Thus compound **1** was identified as 2'- $O$ -acetyldihydropenstemide.

On the basis of decoupling experiments on the  $^1\text{H}$  NMR spectrum of compound **1**, we found that the published shifts for H-6 and H-7 in the  $^1\text{H}$  NMR spectrum of dihydropenstemide [11] should be changed as follows: H-6,  $\delta$ 1.69 and 1.82; H-7,  $\delta$ 1.4 and 1.9–2.2.

Compound **2** had a similar IR spectrum to that of compound **1** (3450, 1730–1750, 1670, 1260  $\text{cm}^{-1}$ ). Acetylation afforded a hexa-acetate (**2a**) with a  $M_r$  of 714 (CIMS with  $\text{Et}_2\text{NH}$ ). The  $^1\text{H}$  NMR spectrum of **2** showed the presence of one acetoxy group and hence a  $M_r$  of 504 for **2**. The MS fragmentation of compound **2** resembled that of the penta-acetate of 2'- $O$ -acetyldihydropenstemide (**1a**) (Scheme 1 and Experimental).

Comparison of the  $^{13}\text{C}$  NMR data of the hexa-acetate of compound **2** and an authentic sample of patrinoside hexa-acetate (**4a**) [4, 14, 15] confirmed the identity of

both compounds (Table 2). As with 2'- $O$ -acetyldihydropenstemide (**1**), the  $^1\text{H}$  NMR data showed that the acetoxy group in compound **2** was not attached at the aglycone, i.e. H-10 and H-7 signals unchanged (Table 1). A 2'-location of the acetoxy group in the glucosidic moiety was determined on the basis of the same considerations as in the case of compound **1** (Tables 1 and 2).

The similar  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra of compound **2** and patrinoside (**4**) proved that the former contained 7- $\beta$ -OH and 8- $\beta$ - $\text{CH}_2\text{OH}$  groups. Compound **2** was thus identified as 2'- $O$ -acetylpatrinoside.

#### EXPERIMENTAL

$^1\text{H}$  NMR (250 MHz) and  $^{13}\text{C}$  NMR (62.9 MHz): solvents as indicated with TMS as int. standard (accuracy  $\pm 0.25$  Hz).

*Extraction and isolation.* Dried powdered bark of *Viburnum lantana* (750 g) from Mount Vitoshka was extracted with 5 l  $\text{CHCl}_3$  and 3  $\times$  5 l MeOH. The MeOH extracts were bulked and the solvent removed to give a residue (42 g) which was dissolved in  $\text{H}_2\text{O}$  and successively extracted with  $\text{Et}_2\text{O}$ , EtOAc and BuOH. The EtOAc extract was concd to dryness to give a residue (7 g) which was applied to a silica gel column. Elution with  $\text{CHCl}_3$ -MeOH (8:1) with increasing MeOH content gave impure compounds **1** and **2**, which were repeatedly chromatographed until pure.

compounds **1**–**3**, **1a**, **2a** and **4a**

<b>2*</b>	<b>2a†</b>	<b>4a§</b>
5.93 <i>d</i> (5.0)	5.92 <i>d</i> (5.3)	5.86 <i>d</i> (6)
6.31 <i>br s</i>	6.33 <i>d</i> (1.5)	6.25 <i>d</i> (1.5)
2.89 <i>m q-habitus</i>	2.88 <i>m q-habitus</i>	2.7–3.2 <i>m</i>
1.80 <i>m</i>	1.9 <i>m</i>	2.2
1.97 <i>m</i>	1.9–2.3 <i>m</i>	
4.31 <i>m</i>	5.27 <i>m</i>	
1.90–2.10 <i>m</i>	1.9–2.3 <i>m</i>	
2.18 <i>m</i>	1.9–2.3 <i>m</i>	
3.83 <i>dd AB</i> from ABX (10.0; 6.0)	4.18 <i>m</i>	4.05–4.20
3.71 <i>dd AB</i> from ABX (10.0; 7.5)		
4.22 <i>d</i> (11.0)	4.20 <i>d AB</i> (11.5)	
4.05 <i>d</i> (11.0)	4.06 <i>d AB</i> (11.5)	
4.46 <i>d</i>	4.52 <i>d</i> (8.0)	
4.71 <i>dd</i> (11.0)	5.02 <i>dd</i> (8.0; 9.5)	
3.52 <i>t</i> (9.0)	5.21 <i>t</i> (9.5)	
3.35	5.05 <i>t</i> (9.5)	
3.27	3.70 <i>m</i>	
3.88 <i>dd AB</i> from ABX (12.0; 2.0)	4.26 <i>dd ABX</i> (11.0; 5.0)	
3.67 <i>dd AB</i> from ABX (12.0; 4.5)	4.15 <i>dd AB</i> (11.0; 2.5) from ABX	
2.10 <i>s</i>		
1.9–2.1 <i>m</i>	1.9–2.3	1.95–2.10
2.24 <i>d</i>	2.27	
0.96 <i>d</i>	0.98 <i>d</i>	0.96 <i>d</i> (6)

Table 2.  $^{13}\text{C}$  NMR data of compounds **1–4** and **1a–4a**

C	<b>1</b> *	<b>3</b> †	<b>1a</b> ‡	<b>3a</b> §	<b>2</b> *	<b>4</b> *	<b>2a</b> ‡	<b>4a</b>
1	92.9	93.2	91.4	91.0	93.5	93.5	91.3	90.8
3	140.7	140.6	140.4	140.0	140.1	139.9	140.1	139.7
4	114.5	115.2	112.2	111.7	116.2	115.7	112.7	112.2
5	36.8	36.9	36.1	35.6	34.0	33.4	33.1	32.7
6	30.6	30.9	30.1	29.5	41.0	39.9	37.5	37.0
7	28.0	28.1	27.4	26.9	73.5	72.3	74.1	73.5
8	44.2	43.9	39.1	38.6	49.0	48.1	43.2	42.7
9	45.0	45.0	43.8	43.3	41.0	41.5	42.2	41.7
10	66.6	66.5	67.3	66.8	62.3	61.5	62.5	62.0
11	69.6	69.6	69.0	68.5	69.8	69.6	68.8	68.8
1'	102.2	103.5	99.3	98.8	101.3	102.0	99.2	98.6
2'	75.4	75.2	71.4	70.9	75.5	73.9	71.4	70.8
3'	76.3	77.9	72.0	71.4	76.3	76.6	72.0	71.5
4'	71.9	71.8	68.0	68.0	71.9	70.4	68.5	68.0
5'	78.1	78.2	73.0	72.5	78.2	76.6	73.0	72.5
6'	62.8	62.9	62.0	61.5	62.8	61.5	62.0	61.5
isovaleroyl								
>C=O	173.4	173.5	171.9		173.5		171.8	
–CH <sub>2</sub> –	44.2	44.2	43.3		44.3		43.2	
–CH<	26.8	26.8	25.6		26.9		25.6	
–Me × 2	22.6	22.6	22.3		22.7		22.3	
acetyl								
>C=O	171.7				172.0			
Me	21.7				21.2			

**1** and **2** in  $\text{CD}_3\text{OD}$ , 62.9 MHz

† **3** in  $\text{CD}_3\text{OD}$ , 100 MHz, ref. 11

‡ **1a** and **2a** in  $\text{CDCl}_3$ , 62.9 MHz

§ **3a** in  $\text{CDCl}_3$ , 22.6 MHz, ref. 4.

\* **4** in  $\text{D}_2\text{O}$ , 22.6 MHz, ref. 14.

|| **4a** in  $\text{CDCl}_3$ , 22.6 MHz, ref. 14.

**2'-O-Acetyldihydropenstemiide (1).** Amorphous powder (94 mg). UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm: 212; IR  $\nu_{\text{max}}^{\text{film}}$   $\text{cm}^{-1}$ : 3450, 1730–1750, 1670, 1375, 1250, 1090, 765;  $^1\text{H}$  NMR (250 MHz,  $\text{CD}_3\text{OD}$ , decoupling experiments), see Table 1;  $^{13}\text{C}$  NMR (62.9 MHz,  $\text{CD}_3\text{OD}$ , DEPT technique), see Table 2; EIMS (70 eV),  $m/z$  (rel. int.): 207 (10), 206 (12), 165 (40)  $[\text{M} - 102 - 221]^+$ , 164 (50)  $[\text{M} - 102 - 222]^+$ , 85 (80), 60 (45), 57 (65), 43 (100); EIMS (12 eV): 249 (11)  $[\text{M} - 222 - 17]^+$ , 207 (20), 205 (22)  $[\text{M} - 102 - 221]^+$ , 165 (85)  $[\text{M} - 102 - 221]^+$ , 164 (55), 85 (70), 60 (30), 57 (60), 43 (20).

**Acetylation of 1.** Compound **1** (52 mg) was treated with pyridine- $\text{Ac}_2\text{O}$  in the usual manner. The resultant acetate was purified by silica gel CC to give 45 mg of the pentaacetate **1a**,  $\text{C}_{31}\text{H}_{44}\text{O}_{15}$ , mp 97–99°. IR  $\nu_{\text{max}}^{\text{CHCl}_3}$   $\text{cm}^{-1}$ : 1750, 1670, 1375, 1250, 1050, 765;  $^1\text{H}$  NMR (250 MHz,  $\text{CDCl}_3$ ), see Table 1.  $^{13}\text{C}$  NMR (62.9 MHz,  $\text{CDCl}_3$ ), see Table 2. CIMS ( $\text{Et}_3\text{NH}$ ),  $m/z$  (rel. int.): 730 (100)  $[\text{M} + 74]^+$ , 422 (60)  $[\text{M} + 74]^+$ , 298 (50)  $[\text{M} - 102 - 330]^+$ ; EIMS (70 eV): 331 (40), 207 (70)  $[\text{M} - 102 - 347]^+$ , 206 (72), 169 (50), 109 (20), 85 (60), 57 (60), 43 (100); EIMS (12 eV): 207 (100), 206 (100), 169 (100), 147 (25), 109 (20), 85 (35), 57 (10), 43 (12).

**Acid hydrolysis of 1.** Compound **1** (5 mg) was dissolved in 1 ml 0.5 M  $\text{H}_2\text{SO}_4$ , and the mixture refluxed for 1 hr. After neutralization and removal by filtration of the resinous products, the  $\text{H}_2\text{O}$  soln was concd. D-Glucose was identified by TLC ( $\text{EtOAc}$ –pyridine– $\text{H}_2\text{O}$  2:1:2) and GC of the silylated derivative.

**2'-O-Acetylpatrinoside (2).** Hydroscopic amorphous powder (97 mg). UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm: 212; IR  $\nu_{\text{max}}^{\text{film}}$   $\text{cm}^{-1}$ : 3450, 1750, 1670, 1375,

1260, 1080, 765;  $^1\text{H}$  NMR (250 MHz,  $\text{CD}_3\text{OD}$ , decoupling experiments), see Table 1;  $^{13}\text{C}$  NMR (62.9 MHz,  $\text{CD}_3\text{OD}$ , DEPT technique for multiplicity), see Table 2. EIMS (70 eV),  $m/z$  (rel. int.): 402 (1)  $[\text{M} - 102]^+$ , 384  $[\text{M} - 102 - 18]^+$ , 342 (18)  $[\text{M} - 102 - 60]^+$ , 181 (20)  $[\text{M} - 102 - 221]^+$ , 180 (20), 85 (80), 60 (48), 57 (54), 43 (100).

**Acetylation of 2.** Compound **2** was acetylated with pyridine- $\text{Ac}_2\text{O}$  and the acetylated product purified by silica gel chromatography to give the hexa-acetate **2a** (67 mg),  $\text{C}_{33}\text{H}_{46}\text{O}_{17}$ , mp 129–131°. IR  $\nu_{\text{max}}^{\text{CHCl}_3}$   $\text{cm}^{-1}$ : 1750, 1665, 1435, 1375, 1250, 1050, 770;  $^1\text{H}$  NMR (250 MHz,  $\text{CDCl}_3$ ), see Table 1.  $^{13}\text{C}$  NMR (62.9 MHz,  $\text{CDCl}_3$ ), see Table 2. CIMS ( $\text{Et}_3\text{NH}$ ),  $m/z$  (rel. int.): 788 (82)  $[\text{M} + 74]^+$ , 422 (100)  $[\text{M} + 74]^+$ , 356 (40)  $[\text{M} - 102 - 330]^+$ ; EIMS (70 eV): 383 (4)  $[\text{M} - 331]^+$ , 331 (20), 273 (10), 265 (1), 264 (10), 169 (70), 153 (20), 144 (20), 109 (25), 85 (70), 57 (50), 43 (100).

**Acid hydrolysis of 2.** 6 mg of compound **2** was refluxed with 1 ml 0.5 M  $\text{H}_2\text{SO}_4$  for 1 hr. After neutralization and removal of the resinous products by filtration, the water phase was concentrated and D-glucose was identified by TLC and GC of the silylated derivative.

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